Solubility Rules

The common solubility rules (in water) are:

- 1. All compounds of Ammonium (NH₄⁺) and Group 1 cations are soluble.
- 2. All nitrates are soluble.
- 3. All chlorides, bromides and iodides are soluble except for silver, lead and mercury.
- 4. All sulfates are soluble except for silver, lead, mercury, barium, strontium and calcium.
- 5. All carbonates and phosphates are insoluble except for ammonium and Group 1cations.
- 6. All hydroxides are insoluble except for ammonium, calcium, barium and Group 1cations.