## Concentration Calculations 2

1. If 4.0 g of Calcium is reacted with 200 mL of a $2 \mathrm{molL}^{-1}$ solution of hydrochloric acid. Calculate the volume of gas produced at 273 K and 100 kPa .
2. If 1.5 grams of calcium carbonate is mixed with 100 mL of a $2.0 \mathrm{molL}^{-1}$ sulfuric acid calculate the volume of gas produced at 298 K and 100 kPa
